

Contents lists available at ScienceDirect

# Sensors and Actuators: B. Chemical

journal homepage: www.elsevier.com/locate/snb





# Label-free and sensitive detection of *Citrus Bark Cracking Viroid* in hop using Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene-modified genosensor

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## ARTICLE INFO

## Keywords: CBCVd Viroid Hop RNA detection Impedimetric genosensor

# ABSTRACT

Since the discovery of Citrus bark cracking viroid (CBCVd) in hops in 2007, affected hop-growing countries such as Slovenia and Germany have been actively pursuing efficient and easily accessible diagnostic tools that could contribute to the early detection of CBCVd in the field. In the early stages of CBCVd infection, typical symptoms or subtle signs of spread may not be evident. Detection becomes feasible only when the plant begins to display symptoms. Unfortunately, there is currently no treatment available, which requires the removal of infected plants as the only viable solution. However, this approach leads to significant economic losses on a large scale. This work demonstrates the development and study of a sensitive, selective, and label-free impedimetric genosensor for the detection of CBCVd in total RNA hop samples. The genosensor is based on a supporting glassy carbon electrode modified with streptavidin-agarose beads, which serve as an effective immobilization layer for a biotinylated single-stranded DNA capture probe. The integration of a 2D-layered Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene into the sensing architecture resulted in a significantly improved electroanalytical performance of the genosensor. Several fabrication and operational parameters were optimized, such as the streptavidin-agarose beads deposition time. capture probe immobilization time and concentration, and sample incubation time. The optimized genosensor exhibited a limit of detection of only 0.5 fg  $\mu$ L<sup>-1</sup> (5.5 fmol L<sup>-1</sup>) in combination with a one-hour incubation with the denatured total RNA hop extract, thus eliminating the need for an additional and laborious amplification step.

## 1. Introduction

The hop ( $Humulus\ lupulus\ L$ .) is an economically important crop due to the worldwide use of its secondary metabolites in breweries [1] and pharmaceutical companies [2]. Over the past three decades, global hop production has surged, with the United States and Germany leading the way [3]. Hops are susceptible to many diseases, including those caused

by viroids, the smallest biological pathogens capable of replication using host enzymes. Viroid infections can be highly aggressive, resulting in reduced plant growth, dwarfing, abnormal leaf shapes, yellowing, necrosis, and leaf rolling [4]. The structure of viroids consists of non-encapsulated, single-stranded, circular RNA molecules ranging in size from 239 to 401 nucleotides, whose nucleotide sequence does not allow protein-coding [5]. Viroids spread through various ways,

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predominantly by mechanical contact with agricultural equipment, by leaf-to-leaf contact, and by human handling, as well as by seeds, pollen, and insect vectors [6].

The Citrus bark cracking viroid (CBCVd) was first discovered in citrus fruits in 1988 and was classified as a mild pathogen for these plants [7, 8]. However, in 2007, CBCVd was discovered in Slovenian hop plants, leading to the complete eradication of over 300 ha of hop fields due to the viroid's aggressive nature [9,10]. CBCVd is particularly devastating for hop cultivation, causing severe symptoms such as plant stunting (reduced growth and overall size compared to healthy plants), root dry rot, and plant dieback three to five years after infection. Despite the Slovenian national eradication program, CBCVd continues to threaten farmers and hop production as an emerging highly aggressive pathogen. In 2019, a CBCVd outbreak appeared in Germany, and the most recent report comes from Brazil in 2023 [11], posing a threat to hop-growing regions in Europe and worldwide [10].

The emergence of such pathogens underlines the urgent need for rapid and straightforward viroid detection methods. Early identification of plant diseases can prevent or limit the spread of emerging infections, ultimately reducing costs and minimizing damage caused by contamination [12]. The insidious nature of CBCVd, characterized by the absence of typical symptoms or the presence of only atypical symptoms during the early stages of infection, highlights the importance of simple and reliable on-site detection for farmers [9].

Typically, the identification of pathogens such as viruses and viroids relies on conventional or advanced molecular biology techniques [13]. These methods tend to be both costly and time-consuming. Examples of such methods include polymerase chain reaction (PCR), reverse transcription quantitative PCR (RT-qPCR), loop-mediated isothermal amplification (LAMP), enzyme-linked immunosorbent assay (ELISA), and CRISPR-Cas-based detection [14].

Alternatively, electrochemistry offers simple and cost-effective approaches for detecting microbial pathogens. Designing biosensors for plant pathogens typically involves the modification of the electrode surface to immobilize biological recognition elements for interaction with the target [15]. While significant progress has been made in the development of electrochemical biosensors for plant pathogens over the past three decades [16], reports specifically addressing viroids remain scarce. For example, a graphene/zinc oxide nanocomposite genosensing platform capable of detecting a CCCVd (Coconut cadang-cadang viroid) RNA oligonucleotide over a linear range of  $1.0 \times 10^{-11}$  M to  $1.0 \times 10^{-6}$  M was presented [17]. The initial modification layer was functionalized with 1-pyrenebutyric acid N-hydroxysuccinimide ester to enable chemical binding of the single-stranded DNA capture probe. Subsequent hybridization with the single-stranded RNA oligonucleotide target was monitored by either electrochemical impedance spectroscopy (EIS) or differential pulse voltammetry (DPV). More recently, a similar genosensor for CCCVd detection was developed based on gold nanoparticles decorated phosphorene nanohybrid with graphene, which achieved a similar linear range of  $1.0 \times 10^{-11}$  M to  $1.0 \times 10^{-6}$  M, with a limit of detection of  $2.8 \times 10^{-12}$  M [18]. In both studies, the detection target consisted of artificially prepared single-stranded RNA corresponding to a segment of the CCCVd sequence.

In this work, a sensitive and selective genosensor was demonstrated, exploiting streptavidin-agarose beads as a biocompatible immobilization matrix for a single-stranded DNA capture probe and the attractive electrocatalytic properties of 2D-layered  $\rm Ti_3C_2T_x$  MXene. The genosensor successfully detected CBCVd in infected total RNA hop samples obtained directly from the field, without the need for labeling or laborious amplification steps.

## 2. Experimental

## 2.1. Chemicals and materials

All reagents were of analytical grade and were used without further

purification. All aqueous solutions were prepared using ultrapure water with resistivity not less than 18.2 M $\Omega$  cm at 298 K (Milli-Q, Millipore, Corp., Marlborough, USA). Titanium aluminum carbide (Ti $_3$ AlC $_2$ ) was acquired from Jinzhou Haixin Metal Materials (China).

Streptavidin-agarose beads (STR-AGAR beads) from *Streptomyces avidinii*, sodium hydroxide (NaOH), and sodium chloride (NaCl) were purchased from Merck (Darmstadt, Germany), sodium azide (NaN<sub>3</sub>), glycine, and bovine serum albumin (BSA) from Sigma Aldrich (St. Louis, USA), disodium hydrogen phosphate dihydrate (Na<sub>2</sub>HPO<sub>4</sub>·2 H<sub>2</sub>O) from Kemika (Zagreb, Croatia), alumina (Al<sub>2</sub>O<sub>3</sub>, 0.3  $\mu$ m) and polishing cloth from Buehler (Lake Bluff, USA), potassium dihydrogen phosphate (KH<sub>2</sub>PO<sub>4</sub>) and potassium hexacyanoferrate (III) (K<sub>3</sub>[Fe(CN)<sub>6</sub>]) from Fluka (Switzerland), and potassium chloride (KCl) and potassium hexacyanoferrate (II) (K<sub>4</sub>[Fe(CN)<sub>6</sub>]) from Riedel-de Haën (Seelze, Germany). TE buffer (10 mM Tris-HCl, 1 mM EDTA, pH 8.0) and nuclease-free water were purchased from Integrated DNA Technologies (Coralville, USA). The phosphate-buffered saline (PBS, pH=7.4) consisted of 8.00 g of NaCl, 200.0 mg of KCl, 1.81 g of Na<sub>2</sub>HPO<sub>4</sub>·2 H<sub>2</sub>O, and 240.0 mg of KH<sub>2</sub>PO<sub>4</sub> in 1.0 L purified H<sub>2</sub>O.

The capture probe sequence, 50 bp in length (5'-AGAGTTGTATC-CACCGGGTAGTTTCTTCTCAGGTCGCGAAGGAAGAAGC-3'), was synthesized by Eurofins Genomics (Germany) with a biotin moiety attached to the 5' end. This probe was designed based on the sequence of the CBCVd isolate Hl-1 (GenBank accession number KM211546.1). The capture probe anneals perfectly to the region spanning nucleotides 182–231 of the CBCVd genome.

## 2.2. Total RNA isolation and CBCVd quantification

The viroid source was infected hop plants of the variety 'Celeia' from the pathogen reference collection of the Diagnostic Laboratory for Plant Protection at the Institute of Hop Research and Brewing, Žalec, Slovenia. The plants were biolistically inoculated in 2013 with infectious cDNA constructs (GenBank X07397, GenBank KM211546, GenBank X07405) of single viroids (CBCVd, HLVd) and combinations (CBCVd+HLVd and HSVd+HLVd) [19]. Viroid-free hop plants of the same variety and tomato (Solanum lycopersicum L.; cv. Heinz 1370) plants served as the source of negative control plant samples. From each plant, three fully developed leaves were sampled at phenological stage BBCH 38 for hops and BBCH 29 for tomatoes, then stored at -80°C until being used for RNA isolation. Total RNA was extracted from 100 mg of leaf tissue taken from all three sampled leaves of each plant using a CTAB protocol without DNase treatment [20]. The total RNA was eluted in  $50 \,\mu\text{L}$  of a TE buffer (10 mM Tris-HCl, 1 mM EDTA, pH 8.0). The RNA concentration was measured using the Qubit® 3.0 Fluorometer (Thermo Fisher Scientific, USA) and stored at -80°C (Table S1).

The High-Capacity cDNA Reverse Transcription Kit (Thermo Fisher Scientific, USA) was used to produce cDNA from total RNA hop samples containing 500 ng RNA. The absolute quantification of CBCVd was performed by 7500 Real-Time PCR System (Thermo Fisher Scientific, USA) using a linearised plasmid containing CBCVd sequence as standards. The reaction was performed using SybrGreen chemistry and primers CBCVd-PCR-F-q: 5'-TCACTGGCGTCCAGCACC-3' and CBCVd-PCR-R-q: 5'-AGGAAGAAGCGACGATCGG-3' [21].

# 2.3. Synthesis of $Ti_3C_2T_x$ MXene

The synthesis of  ${\rm Ti_3C_2T_x}$  MXene followed the same protocol as previously published [22]. Briefly, the precursor  ${\rm Ti_3AlC_2}$  MAX phase was etched with 40 % HF for four days, followed by LiF/HCl for two days. The product was then repeatedly washed and centrifuged with H<sub>2</sub>O until approaching neutral pH, being dried at 50°C in a vacuum oven. The resulting material was then stored in polar organic solvent at 4°C.

# 2.4. Characterization of $Ti_3C_2T_x$ MXene

The morphology of the precursor and synthesized materials was examined by scanning electron microscopy (SEM) with field emission Tescan LYRA3 GM dual-beam microscope and Tescan MAIA3 (Tescan, Czech Republic) at an acceleration voltage of 5 kV. Elemental composition was investigated by energy-dispersive X-ray spectroscopy (EDX) using an X-MaxN detector from Oxford Instruments at 20 kV acceleration voltage. For both SEM and EDX analyses, the Ti<sub>3</sub>AlC<sub>2</sub> powder and the colloidal suspension of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene were drop-cast on copper tape. Scanning transmission electron microscopy (STEM) was performed at Tescan MAIA3 electronic microscope with a STEM sample holder and X-MaxN 150 EDX detector. The diluted suspension of MXene was dropcast on a 200 mesh Cu TEM grid (Ted Pella, USA), and then dried. STEM measurements were carried out using a 30 kV electron beam. An InVia Raman microscope (Renishaw, England) was used for Raman spectroscopy measurements in backscattering geometry with a CCD detector. An Nd:YAG laser (532 nm, 50 mW) with a 2400 line mm<sup>-1</sup> diffraction grating, an applied power of 5 %, and a 20x objective were used for the measurements. The Raman spectrum was collected with 100 accumulations in the Raman range of 100–900 cm<sup>-1</sup> at room temperature. X-ray diffraction (XRD) measurements were carried out using a Bruker D2 Phaser benchtop diffractometer (Bruker, Germany) with a Cu Kα radiation source ( $\lambda$ =1.54178 Å). The diffraction patterns were acquired in the range of  $2\theta$  from 5 to  $80^\circ$  at room temperature. The data were evaluated by HighScore Plus 4.9 software. Fourier transform-infrared spectroscopy (FTIR) measurements were performed on a Nicolet iS50R FTIR spectrometer (Thermo Scientific, USA). The measurement was performed with a DLaTGS detector and a KBr beam splitter in the range  $4000-400 \text{ cm}^{-1}$  at a resolution of 4 cm<sup>-1</sup>, with a total of 265 scans. FTIR spectra were measured in a KBr pellet made from 300 mg of KBr and 1 mg of materials. OMNIC software was used to acquire and process the data. Baseline correction and data treatment were performed according to recommendations by T. Parker et al. [23].

# 2.5. Genosensor fabrication

Before use, the GCE underwent mechanical polishing using 0.3  $\mu m$  alumina slurry. The GCE was then sonicated in ethanol for 2 min, washed with ultrapure water, and dried using high-purity nitrogen gas (0.2 bar). Next, 40 mg of commercial STR-AGAR beads were dissolved in 2 mL of phosphate-buffered saline (PBS, pH 7.4) containing 0.03 % sodium azide. 10  $\mu L$  of the STR-AGAR beads solution was carefully drop-casted onto the GCE surface and stored at 4 °C for 2 h.

Modification with  $Ti_3C_2T_x$  MXene: the electrocatalytic 2D-layered  $Ti_3C_2T_x$  MXene in ethanol was heated to approximately  $30^{\circ}C$  until the solvent evaporated slowly. The remaining powder was then redispersed in ultrapure water to obtain an aqueous solution with a concentration of 2 mg mL $^{-1}$ . After vortexing for 1 min, the MXene suspension was mixed with the STR-AGAR beads solution in a 1:1 ratio. The resulting  $Ti_3C_2T_{x^-}$  STR-AGAR mixture was vortexed for 10 seconds before depositing 4.5  $\mu$ L of it onto the supporting GCE surface and stored at 4 °C for 2 h.

Between each modification step, the electrode surface was rinsed with ultrapure water for 1 min and dried using a gentle stream of nitrogen gas (0.2 bar).

Next, 8  $\mu$ L of the biotinylated capture probe solution (50 ng  $\mu$ L<sup>-1</sup>) was drop-casted onto the modified electrode surface and incubated for 2 h at 4 °C. To prevent non-specific binding, 8  $\mu$ L of glycine solution (10  $\mu$ g mL<sup>-1</sup>) was deposited onto the surface of the genosensor for 30 min at 4 °C.

## 2.6. EIS genosensing

Total RNA hop samples extracted from hop plants underwent denaturation in a 95°C water bath for 3 min, followed by rapid cooling on ice for 1 min. Finally, 8  $\mu L$  of the real sample solution was applied to the

sensor surface and incubated for 60 min at room temperature. The sensor was then quickly rinsed with ultrapure water, and electrochemical detection was conducted in the presence of 5 mM [Fe(CN) $_6$ ]<sup>3-/4</sup> in 0.1 M KCl. The fabrication of the genosensor and detection of CBCVd are depicted in Fig. 1.

## 2.7. Electrochemical measurements

Cyclic voltammetric (CV) and electrochemical impedance spectroscopic (EIS) measurements were carried out with a portable PalmSens 4 electrochemical workstation (potentiostat/galvanostat) operated by the PSTrace 5.9 software (PalmSens BV, Netherlands). In all experiments, a glassy carbon supporting electrode (GCE, 3 mm) was used, in combination with a single junction Ag/AgCl/KCl(satd.) reference electrode, and a Pt rod as the counter electrode. All electrodes were purchased from Metrohm (Herisau, Switzerland). EIS spectra were recorded in the frequency range of  $10^5$ - $10^{-1}$  Hz, at the potential of +0.14 V, with an amplitude of 5 mV, in a solution of 5 mM [Fe(CN)<sub>6</sub>]<sup>3-/4-</sup> in 0.1 M KCl. CV measurements were performed in the potential range of -0.5–1.0 V or -0.2–0.6 V, with a scan rate of 100 mV s<sup>-1</sup>.

#### 3. Results and discussion

# 3.1. Optimization of the genosensor fabrication and operation

The genosensor development protocol was optimized with respect to several parameters. The increase in charge transfer resistance ( $R_{ct}$ ) of the genosensor before and after incubation with a total RNA hop sample infected with CBCVd was used as a feedback signal to determine the optimal fabrication and measurement parameters, i.e.,  $\Delta R = R_{inc} - R_{blank}$ . All  $R_{\rm ct}$  values, i.e.,  $R_{\rm inc}$  (after sample incubation) and  $R_{\rm blank}$  (before sample incubation), were determined by fitting the raw impedance spectra with a Randles-type equivalent electrical circuit ( $R_S([R_{ct}Z_W]$  $Q_{
m dl}$ )) representing the real physicochemical processes at the genosensor/ electrolyte interface. Herein,  $R_{\rm S}$  corresponds to the solution resistance,  $Q_{\rm dl}$  is the constant phase element (CPE) accounting for the non-ideal double-layer capacitance, R<sub>ct</sub> is the charge transfer resistance of the genosensor/electrolyte interface, and  $Z_W$  is the Warburg element that models the diffusion phenomenon. Firstly, we optimized several key parameters influencing the performances of the CBCVd genosensor, i.e., (i) STR-AGAR beads deposition time, (ii) capture probe immobilization time, (iii) sample incubation time, and (iv) capture probe concentration.

The interaction between streptavidin, a tetrameric protein from Streptomyces avidinii, and biotin, a water-soluble vitamin, is one of the fundamental and widespread phenomena in molecular biology and biotechnology [24,25]. This interaction has found numerous applications, including affinity purification [26], immunoassays [27], and biosensing [28]. In addition to a strong immobilization of the biotinylated single-stranded DNA capture probe, the STR-AGAR beads serve as an effective adhesion agent to prevent leaching of the sensing layer from the electrode surface and its dissolution during exposures to solutions in the subsequent fabrication steps. The STR-AGAR beads were introduced from the 20 mg mL<sup>-1</sup> PBS solution (pH 7.4) by a simple drop-casting procedure at 4°C. As depicted in Fig. 2a, the highest analytical signal of the genosensor for 720 fg  $\mu L^{-1}$  CBCVd was obtained when STR-AGAR beads were left to deposit for two hours, whereas shorter and longer deposition times resulted in lower signals. After the optimization of the STR-AGAR beads deposition time, we investigated the effect of the biotinylated capture probe immobilization time upon the genosensor impedimetric response. It was observed (Fig. 2b) that the two-hour immobilization exhibited the highest performance of the genosensor, considering both the signal height and reproducibility when compared to the shorter or longer exposure times to the biotinylated capture probe. The binding event between the STR units confined to the electrode surface and the biotinylated capture probe moieties is a dynamic process that also involves the reverse reaction; however, at the

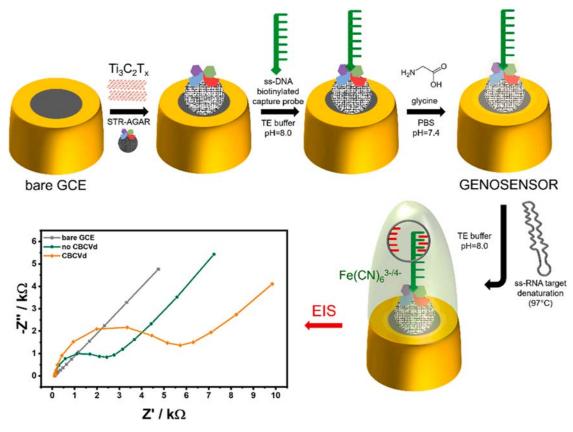


Fig. 1. Fabrication of the CBCVd genosensor and its impedimetric detection.

same time, the steric effects should be taken into consideration [25]. Consequently, a one-hour immobilization might be insufficient to stabilize the sensing surface with the immobilized capture probe; in contrast, longer immobilization times than two hours promote the reverse reaction, i.e., detachment of the capture probe, and/or its improper conformation that leads to poorer performance. Finally, we optimized the time necessary for incubating the total RNA hop sample, and the corresponding results are shown in Fig. 2c. As can be seen, shorter incubation times of 15 and 30 min led to significantly lower signals in comparison with incubation times of 60 and 90 min. Notably, the incubation time of 60 min revealed the highest impedimetric signal along with favorably improved reproducibility. In continuation, all experiments in this study were carried out using the incubation time of one hour.

The next study tackled the effect of capture probe concentration on the genosensor performance, as shown in Fig. S1. Evidently, the impedimetric response was the highest when the biotinylated capture probe concentration was 50 ng  $\mu L^{-1}$ , along with its favorable reproducibility; this concentration was used for all subsequent studies.

# 3.2. Electrochemical characterization of the genosensor fabrication

Using previously optimized parameters, a step-by-step investigation of the genosensor fabrication process was conducted in combination with CV and EIS detection in 5 mM [Fe(CN) $_6$ ]  $^{3\cdot/4-}+0.1$  M KCl, including the final incubation step in total RNA hop samples. By following the electrochemical signals, we gained insights and control into gradual changes occurring at the genosensor surface upon each

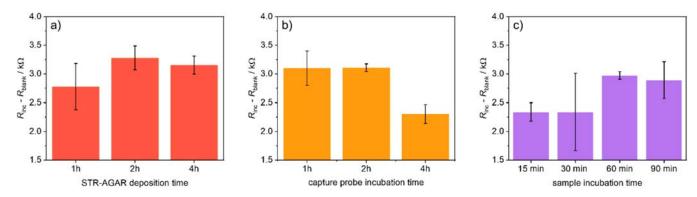


Fig. 2. The effect of (a) STR-AGAR beads deposition time (capture probe incubation time: 2 h, sample incubation time: 60 min), (b) capture probe incubation time (STR-AGAR beads deposition time: 2 h, sample incubation time: 2 h, capture probe incubation time: 2 h) upon the genosensor response. The concentration of CBCVd was 720 fg  $\mu$ L<sup>-1</sup>, and the concentration of the capture probe was 25 ng  $\mu$ L<sup>-1</sup>. The error bars represent the relative standard deviation of three independent measurements. Relevant EIS datasets corresponding to the optimization studies are presented in Table S2.

modification step, attaining feedback to successfully fabricate a genosensor with optimized sensitivity and reproducibility. Using a simple Randles circuit  $R_S([R_{ct}Z_W]Q_{dl})$ , we quantified the impedimetric changes occurring at the genosensor/electrolyte interface. The bare GCE exhibited an impedimetric behavior with a negligible high-frequency semicircle and a diffusion tail typical for conductive carbon surfaces (Fig. 3a). After the introduction of STR-AGAR beads, the impedimetric spectrum altered with the formation of a semicircle, corresponding to a significant increase in the charge-transfer resistance due to the deposition of a less conductive coating. In addition, the incorporation of a biotinylated capture probe further increased the size of the semicircle, reflecting an increased Rct; similarly, the last modification layer, including glycine, contributed to a significantly enlarged R<sub>ct</sub>, as shown in Fig. 3a. Finally, the dark-green EIS spectrum shows the genosensor response after incubation in a real total RNA hop sample solution of 250 fg μL<sup>-1</sup> CBCVd, revealing a well-developed impedimetric signal. Similarly, the cyclic voltammograms (Fig. 3b) showed a stepwise decrease of the anodic and cathodic signals of [Fe(CN)<sub>6</sub>]<sup>3-/4-</sup> accompanied by increasing peak-to-peak separation that corresponds to the deposition of genosensor building blocks, reflecting the changes of the interface diffusion pattern. A small signal observed at bare GCE (at ca. -0.4 V) can be attributed to the oxygen reduction reaction. At the same time, Fig. 3 demonstrates a significantly enhanced sensitivity of the EIS vs. CV, particularly considering the detection of the CBCVd viroid in the real sample (dark-green curve).

To test both the potential influence of the supporting buffer and the stability of the sensing layer, 8  $\mu L$  of the TE buffer was drop-casted onto the fully fabricated genosensor, followed by a one-hour incubation step. As shown in Figure S2, the incubation for one hour in the TE buffer did not affect the EIS response of the genosensor; this was also supported by the fitting parameters, indicating negligible changes, as shown in Table S4.

The incorporation of the  $Ti_3C_2Tx$  MXene into the genosensor architecture unveiled a significant improvement in the electrochemical communication between the upper genosensor interface, where hybridization and detection take place, and the lower part, which includes the STR-AGAR beads combined with  $Ti_3C_2Tx$  MXene and the supporting GCE. MXenes are state-of-the-art 2D-layered materials composed of transition metal carbides or nitrides, which exhibit remarkable versatility in various application areas, such as catalysis [29,30], ion batteries [31,32], gas sensors [22,33], and biosensors [34,35]. Upon introducing  $Ti_3C_2T_x$ , MXene, the preliminary measurements revealed a ca. 5-fold increase in the sensitivity of the genosensor for detecting CBCVd in

real total RNA hop samples. The step-by-step fabrication of the  ${\rm Ti}_3 C_2 T_x$  MXene-modified genosensor is shown in Figure S3, with the corresponding fitting parameters listed in Table S5.

## 3.3. Structural and surface characterization of $Ti_3C_2T_x$ MXene

The MXene Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> was obtained by selective etching of the Al layer of the precursor Ti<sub>3</sub>AlC<sub>2</sub> MAX phase, as described in the Experimental section. The SEM and STEM micrographs of the observed flakes displayed the typical accordion-like morphology of the multilayer Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene, as shown in Fig. 4a and S4. The flakes exhibit tightly packed and loosely packed features depending on the extension in which Al atoms are replaced by surface functional groups such as oxygen (-O), hydroxyl (-OH), or fluorine (-F) [36]. Subsequently, the interlayer forces become weaker, making their delamination possible. The EDX mapping of elements of the detailed area of the selected Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> flake with loosely packed sheets is shown in Fig. 4b, which shows the abundant distribution of C, O, F, and Ti elements, while Al is scarcely detected. Another example of SEM/EDX mapping for  $\text{Ti}_3\text{AlC}_2$  and  $\text{Ti}_3\text{C}_2\text{T}_x$  is shown in Figure S4. The averaged EDX spectra of Ti<sub>3</sub>AlC<sub>2</sub> and Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>, obtained on multiple flakes at 20 kV, are shown in Fig. 4c, and the respective quantification in Table S6. In both cases, the quantified at% of Ti/C indicates a ratio of 1.3, close to the expected for Ti<sub>3</sub>C<sub>2</sub> stoichiometry of 1.5. One of the major differences between the EDX spectra of the materials is the significant decrease of the Al Kα1 line at 1.4 keV, from an initial 17.3 at% to a considerably lower amount of 1.7 at%, also indicating the successful etching of the precursor Ti<sub>3</sub>AlC<sub>2</sub> MAX phase. Furthermore, for Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>, a new peak is quantified at 0.68 keV corresponding to the F Kα line at a level of 26.8 at% for F. The level of quantified O also increased by 20 % for the Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> after the etching of the MAX phase. Both levels of F and O elements are related to the surface functional groups mentioned before.

The Raman spectrum of the precursor MAX phase and MXene obtained in the  $100\text{--}900~\text{cm}^{-1}$  range, are shown in Fig. 4d, providing information related to the bonds present in the structure by assessing lattice vibrations. For the starting  $\text{Ti}_3\text{AlC}_2$ , the peaks at 148, 244, and 434  $\text{cm}^{-1}$  correspond to the  $\text{E}_2$  g,  $\text{E}_1$  g, and  $\text{A}_1$  g vibrational modes, respectively [37]. The peak at 608  $\text{cm}^{-1}$  can be assigned to  $\text{E}_1$  g vibration modes corresponding to the in-plane vibration of the Ti–C bond [38]. In comparison, for the  $\text{Ti}_3\text{C}_2\text{T}_x$  MXene, with the Al layer etching, surface functional groups (=O, -OH, and -F terminations) are adsorbed and intercalated within the interlayer spacing, causing variations in the local environment of the atoms and affecting lattice vibrations of the unit cell.

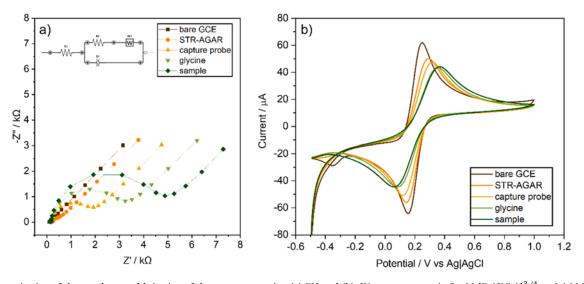


Fig. 3. Characterization of the step-by-step fabrication of the genosensor using (a) EIS and (b) CV measurements in 5 mM  $[Fe(CN)_6]^{3-/4-} + 0.1$  M KCl. The corresponding fitting parameters are listed in Table S3.

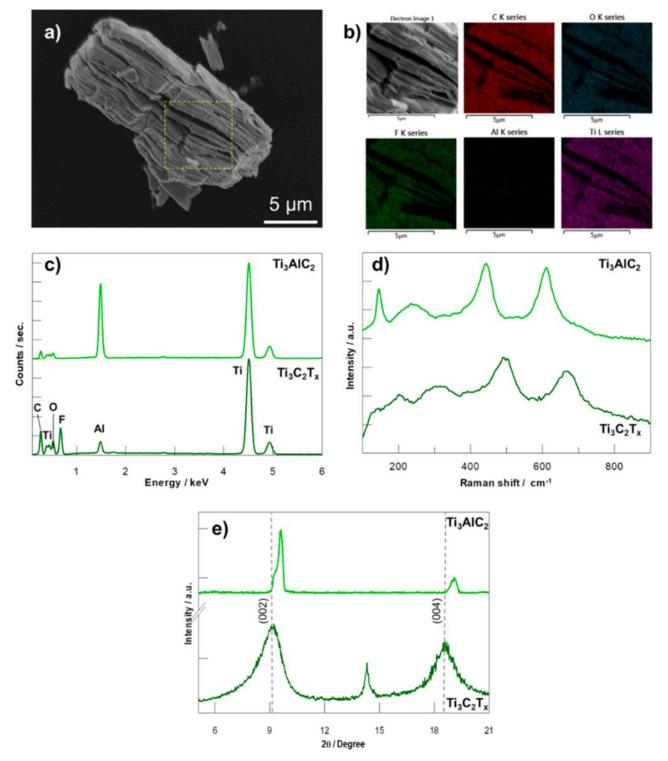


Fig. 4. (a) Overlay of in-beam BSE and SE modes micrograph of a  $Ti_3C_2T_x$  MXene multilayer flake after etching with the highlight of the area used for (b) EDX mapping of elements, (c) averaged EDX spectra of  $Ti_3AlC_2$  MAX phase and  $Ti_3C_2T_x$  MXene, and (d) comparison of Raman spectrum of  $Ti_3AlC_2$  MAX phase and  $Ti_3C_2T_x$  MXene with 532 nm LL. (e) X-ray diffraction (XRD) patterns for low  $2\theta$  region with vertical lines marking the position of (002) and (004) peaks of ICDD card no. 04–14–5068.

Thus, the  $Ti_3C_2T_x$  spectrum peaks have broadened and a blue shift to higher Raman frequencies. The peak at  $148~\rm cm^{-1}$  was initially intense for  $Ti_3AlC_2$ , displaying in-plane vibration Ti atoms, with the major contribution of Ti-Al vibrations, for the case of  $Ti_3C_2T_x$  is greatly decreased, broadened, and shifted to  $201~\rm cm^{-1}$ . The remaining peaks of MXene have higher Raman shifts but peak intensities were sustained, with the peak at  $314~\rm cm^{-1}$  having contributions from out-of-plane

vibration of the Ti–C bond and H atoms of  $Ti_3C_2(OH)_2$  [39]. The peak at 492 cm<sup>-1</sup> has contributions of in-plane vibration of O atoms of OH MXene terminations [40].

X-ray diffraction (XRD) was used to analyze the crystal structure and phase composition of both materials and further understanding of the successful synthesis of  ${\rm Ti}_3{\rm C}_2{\rm T}_x$ . As shown in Figure S5a and Fig. 4e, there is a good match of the diffraction pattern of the precursor  ${\rm Ti}_3{\rm AlC}_2$  and

the ICDD card no. 04–14–5068 [41], displaying a well-crystallized structure, corresponding to layered ternary 312 MAX phase, with a hexagonal crystal system and P6<sub>3</sub>/mmc space group. In Fig. 4e, for the region  $5^{\circ}$ <20<21°, the MAX phase presents the typical diffraction peaks mainly at  $2\theta$ =9.6° and  $19.1^{\circ}$  for (002) and (004) planes, respectively. There was a shift to lower  $2\theta$  values for peaks (002) and (004),  $9.1^{\circ}$  and  $18.6^{\circ}$  respectively for  $Ti_3C_2T_x$ . This is caused by the etching procedure in the synthesis of  $Ti_3C_2T_x$ , removing the Al layers in the initial structure of the MAX phase and induction of new surface terminations in  $Ti_3C_2T_x$  ( $T_x$ =- $F_x$ -0, -OH) [37,42].

As shown in Figure S5b, the FTIR spectrum of the materials exhibits two distinct regions: a 4000-1400 cm<sup>-1</sup> region dominated by peaks associated with H<sub>2</sub>O and C bond vibrations, and a 1400–400 cm<sup>-1</sup> region characterized by chemical fingerprint peaks, related to C-F, Ti-F, Ti-O, and Ti-C vibrations [23]. For both materials, within the 4000–1400 cm<sup>-1</sup> region, the presence of adsorbed H<sub>2</sub>O is evidenced by the O-H stretching at 3600–3200 cm<sup>-1</sup>. Additionally, the O-H bending at 1500–1300 cm<sup>-1</sup> indicates the presence of -OH surface terminations. The C- bond, including C-H stretching, C=O stretching, C-O stretching, and C-H bending, are also observed in this region. For Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>, in the 1400–400 cm<sup>-1</sup> region, the C-F stretching within 1400–1000 cm<sup>-1</sup> is indicative of F surface terminations at defect sites and flake edges, and Ti-C stretching at 464 cm<sup>-1</sup> was also observed in this region. Within the fingerprint region, the Ti-O bending vibrations are the most intense at 560 cm<sup>-1</sup>, typically at 650-550 cm<sup>-1</sup> range. The Ti-F bending, with a typical position of 1000-850 cm<sup>-1</sup>, is prominent at 856 cm<sup>-1</sup>. Some shifts seemed to occur due to the simultaneous presence of Ti, O, and F on the surface of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>, and the presence of confined H<sub>2</sub>O influencing the vibration fingerprint. The C-O stretching exhibits a blue shift from its regular position of 1300–1100 cm<sup>-1</sup> to the 1622 cm<sup>-1</sup> range [23]. These results are also consistent with previous XPS analysis of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> [22], in which the Ti 2p core level was deconvoluted with Ti  $2p_{3/2}$  and Ti  $2p_{1/2}$ peaks assigned to the presence of Ti(IV), Ti(III), and Ti(II) oxidation states. The peaks for the Ti(IV) oxidation state were more intense compared to the other oxidation states, implying a relatively higher surface atomic concentration of Ti(IV)-containing species.

The as-prepared  $Ti_3C_2T_x$  suspension contains inherent polydispersity of the flakes, with a wide range of lateral sizes, as shown in Figures S4g and S4h, which yields an average of 2.05  $\mu$ m. The Tyndall scattering effect by irradiation of a green laser beam through the vial of diluted 0.5 mg mL<sup>-1</sup> MXene dispersion in water is clearly visible, as shown in the inset of Figure S4h. The above results confirm the successful etching of the MAX phase and synthesis of  $Ti_3C_2T_x$  that was subsequently used in the genosensor application.

## 3.4. Analytical performance of the genosensor in real samples

Once the genosensor was fully assembled, further studies were carried out to assess its electroanalytical performance for detecting low concentrations of CBCVd in real samples (Table S1). Notably, we selected the concentration range of CBCVd, mirroring the actual expected concentrations of field-infected plants, by diluting a real total RNA hop sample of a known high concentration of CBCVd.

The performance of the genosensor was investigated by following its impedimetric response when increasing the concentration of CBCVd in a very low concentration range of 0.78–100 fg  $\mu L^{-1}$  (Fig. 5). The genosensor exhibited a highly linear operation in the examined concentration range with an  $r^2$  of 0.99; the calibration curve is characterized by small standard errors for lower CBCVd concentrations and a somewhat large error bar at the highest examined concentration. This can be attributed to the saturation of the sensing surface with the analyte. In contrast, at lower concentrations, the genosensor exhibited higher reproducibility that can be attributed to the mitigated matrix effect due to dilutions of total RNA hop samples, as the real samples utilized in this study exhibited a high degree of CBCVd infection, necessitating their dilutions.

A very low limit of detection (3σ criterium) was achieved, i.e., 0.5 fg

 $\mu L^{-1}$  (5.5 fmol  $L^{-1})$  CBCVd. Such electroanalytical characteristics can be considered remarkable performance for a label-free sensing approach, obviating the need for an additional and time-consuming amplification step.

#### 3.5. Reproducibility and selectivity studies

The genosensor showed good intra-day repeatability and satisfactory inter-day reproducibility. For the intra-day repeatability (n = 3), a value of 12 % was obtained at a CBCVd concentration of 6.25 fg  $\mu L^{-1}.$  Regarding inter-day reproducibility (n = 5), a value of 14.9 % was observed at a concentration of 1.56 fg  $\mu L^{-1}.$ 

Finally, the selectivity study was conducted by performing EIS measurements with the newly developed genosensor incubated with solutions of (i) total RNA hop sample infected with CBCVd, (ii) viroid-free total RNA hop sample, (iii) viroid-free total RNA tomato sample, and (iv) total RNA hop sample infected with *Hop latent viroid* (HLVd). All samples had total RNA concentration of 5 ng  $\mu L^{-1}$ , while the CBCVd concentration in the CBCVd-infected sample was 6.25 fg  $\mu L^{-1}$ . In addition to a very high sensitivity, the genosensor showed a favorable selectivity, as depicted in Fig. 6.

When testing a non-infected total RNA tomato sample and a total RNA hop sample infected with HLVd, both yielded low signals compared to the CBCVd-infected hop sample with the same total RNA concentration. In contrast, a viroid-free total RNA hop sample exhibited even a negative impedimetric response, which could be due to the undefined matrix effect. Such genosensor characteristics hold great promise for its real field application to effectively monitor and control CBCVd infection in hop plants.

# 4. Conclusions

A sensitive and label-free genosensor for the impedimetric detection of CBCVd with good selectivity was developed. The architecture of the genosensor involves non-covalent immobilization of a biotinylated single-stranded DNA capture probe on the supporting GCE modified with a layer of streptavidin-agarose beads with incorporated 2D Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene electrocatalytic material. The 2D-layered  $\text{Ti}_3\text{C}_2\text{T}_x$  was obtained by etching the Al layer of Ti<sub>3</sub>AlC<sub>2</sub> by the HF method, with EDX spectra indicating a significant reduction of Al at% and high levels of F at%. Furthermore, XRD results indicated shifts in (002) and (004) peaks to lower 20 values, caused by removing the Al layers in the initial structure of the MAX phase and induction of new surface terminations in Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>. After optimizing the fabrication and operational parameters, the synergy of the electrocatalytic MXene with other building blocks of the genosensor resulted in a low limit of detection of only 0.5 fg  $\mu L^{-1}$ , positioning this genosensor as a serious candidate for its on-site applications to monitor occurrence and progression of CBCVd directly in hop fields.

# CRediT authorship contribution statement

Alnilan Lobato: Writing – original draft, Visualization, Methodology, Investigation, Formal analysis, Data curation, Conceptualization. Ivan Konjević: Visualization, Investigation, Formal analysis. Sebastjan Radišek: Supervision, Investigation, Funding acquisition, Formal analysis, Data curation. Jernej Jakše: Supervision, Investigation, Funding acquisition, Formal analysis. Helena Volk: Methodology, Investigation, Formal analysis, Data curation. Monika Hermanová: Methodology, Formal analysis. Miroslav Fojta: Supervision, Methodology, Funding acquisition, Formal analysis. Jan Paštika: Investigation, Formal analysis, Data curation. Zdeněk Sofer: Methodology, Investigation, Funding acquisition. Rui Gusmão: Writing – review & editing, Methodology, Investigation, Formal analysis, Data curation. Samo B. Hočevar: Writing – review & editing, Writing – original draft, Supervision, Funding acquisition. Nikola Tasić: Writing – review & editing, Writing

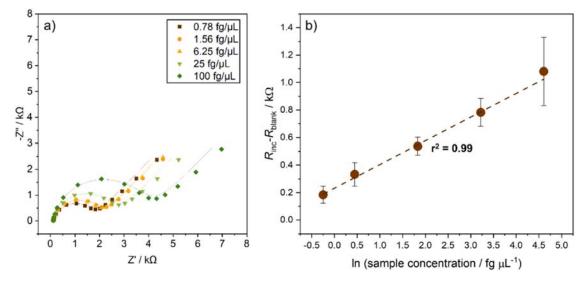
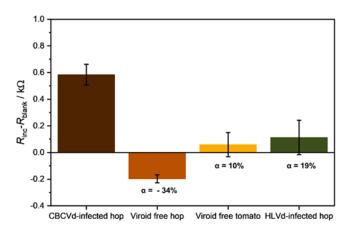


Fig. 5. (a) EIS spectra for increasing concentrations of CBCVd in real total RNA hop samples, and (b) corresponding calibration curve. The error bars represent the relative standard deviation of three independent measurements.



**Fig. 6.** Response of the genosensor in real total RNA samples containing interferents with corresponding selectivity factor values  $(\alpha)$  calculated by dividing the analytical signal of an individual interferent by the analytical signal of the infected total RNA hop sample.

– original draft, Visualization, Supervision, Methodology, Formal analysis, Data curation.

# **Declaration of Competing Interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

# Acknowledgments

This research received funding from the Slovenian Research Agency: Research Program P1–0034, Program P4–0077, Research Project J1–3017, the Slovenian Research Agency's Young Researchers Program - grant agreement No. 56119, and the Czech Science Foundation project 21–46325L. The project was co-financed by the Republic of Slovenia, the Ministry of Education, Science and Sport, and the European Union under the European Regional Development Fund. Z.S. was supported by ERC-CZ program (project LL2101) from Ministry of Education Youth and Sports (MEYS) and AMULET project from Programme Johannes Amos Comenius, co-funded by EU (CZ.02.01.01/00/22\_008/0004558). This work was in part supported from the grant of UCT Prague specific

university research - grant No A2\_FCHT\_2024\_076.

## Appendix A. Supporting information

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.snb.2024.136762.

#### **Data Availability**

Data will be made available on request. Characterization datasets generated and/or analyzed during the study are accessible via the Zenodo repository: 10.5281/zenodo.13889765.

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