Supplementary material

Biocompatible Polyelectrolyte Multilayers with Copper Oxide and Zinc Oxide Nanoparticles for Inhibiting Bacterial Growth

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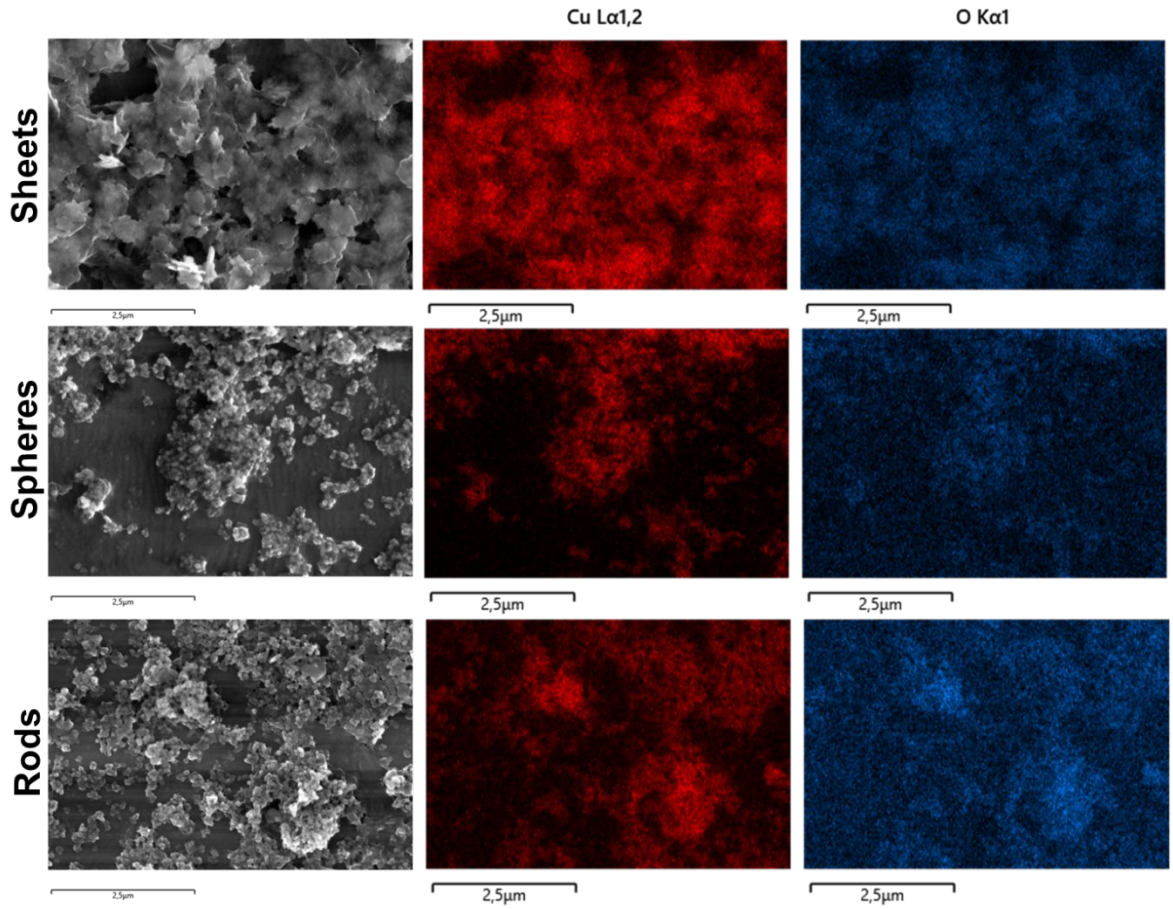

Figure S1. SEM images of stainless-steel surfaces coated with PAH/ALG multilayers and CuO NPs. EDS mapping (second and third column) showing the placement of CuO NPs on the surface. PAH − poly(allylamine hydrochloride), ALG – alginate

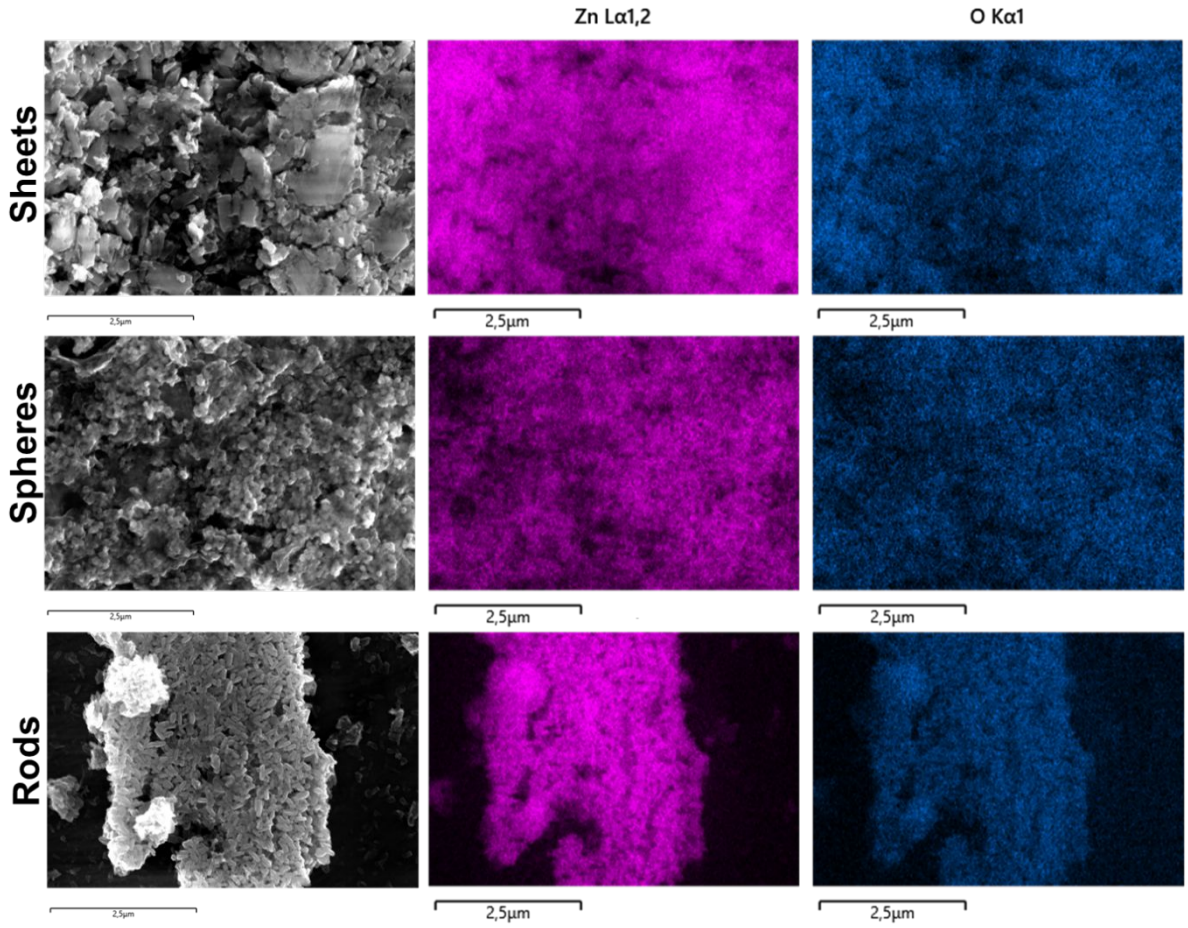

Figure S2. SEM images of stainless-steel surfaces coated with PAH/ALG multilayers and ZnO NPs. EDS mapping (second and third column) showing the placement of ZnO NPs on the surface. PAH − poly(allylamine hydrochloride), ALG – alginate

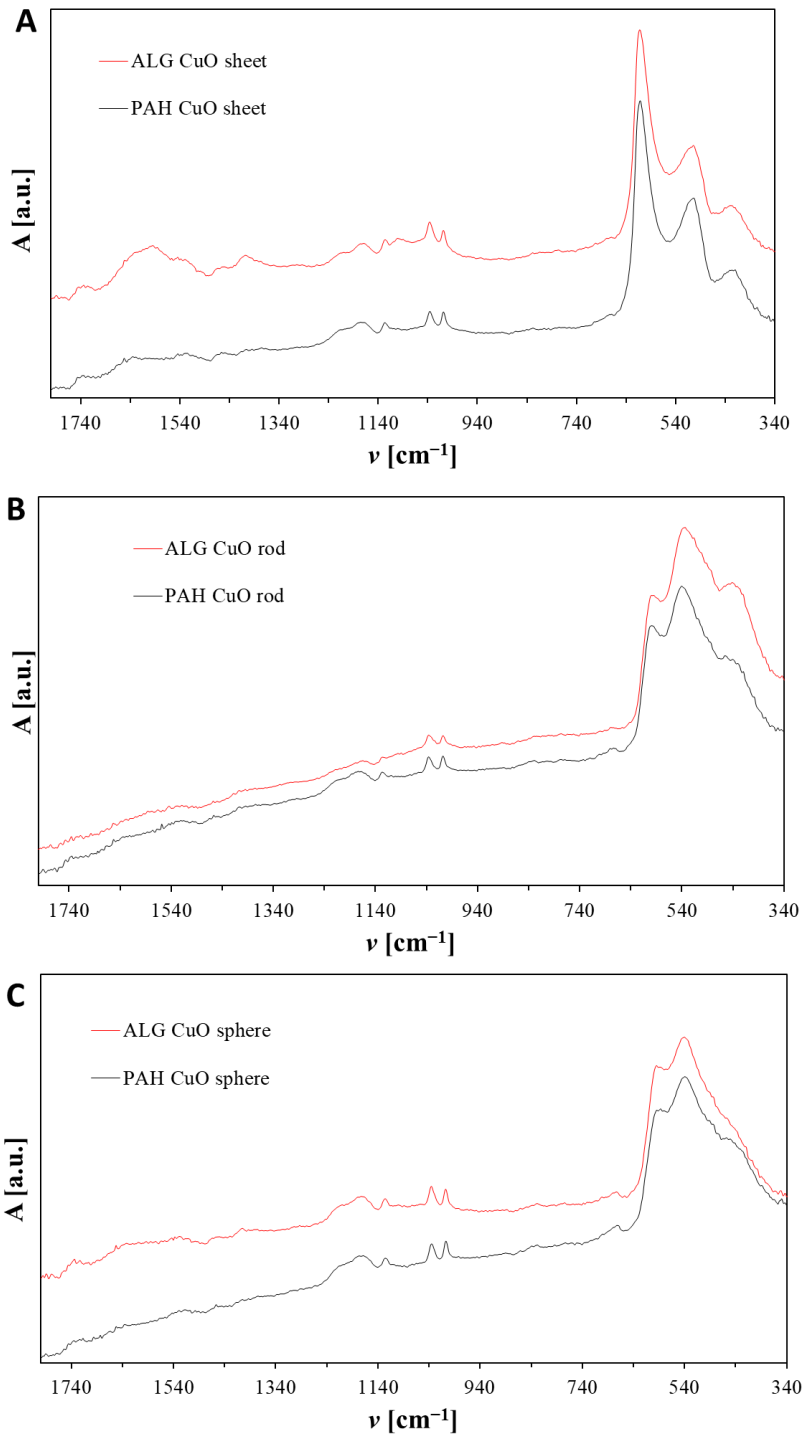

Figure S3. Fourier transformed infrared spectra showing the absorbance (A) at specific wavenumber (ν / cm^{-1}) for PAH and ALG terminating multilayers containing CuO NPs exhibiting different morphology: (A) sheets, (B) rods and (C) spheres. PAH – poly(allylamine hydrochloride), ALG - alginate

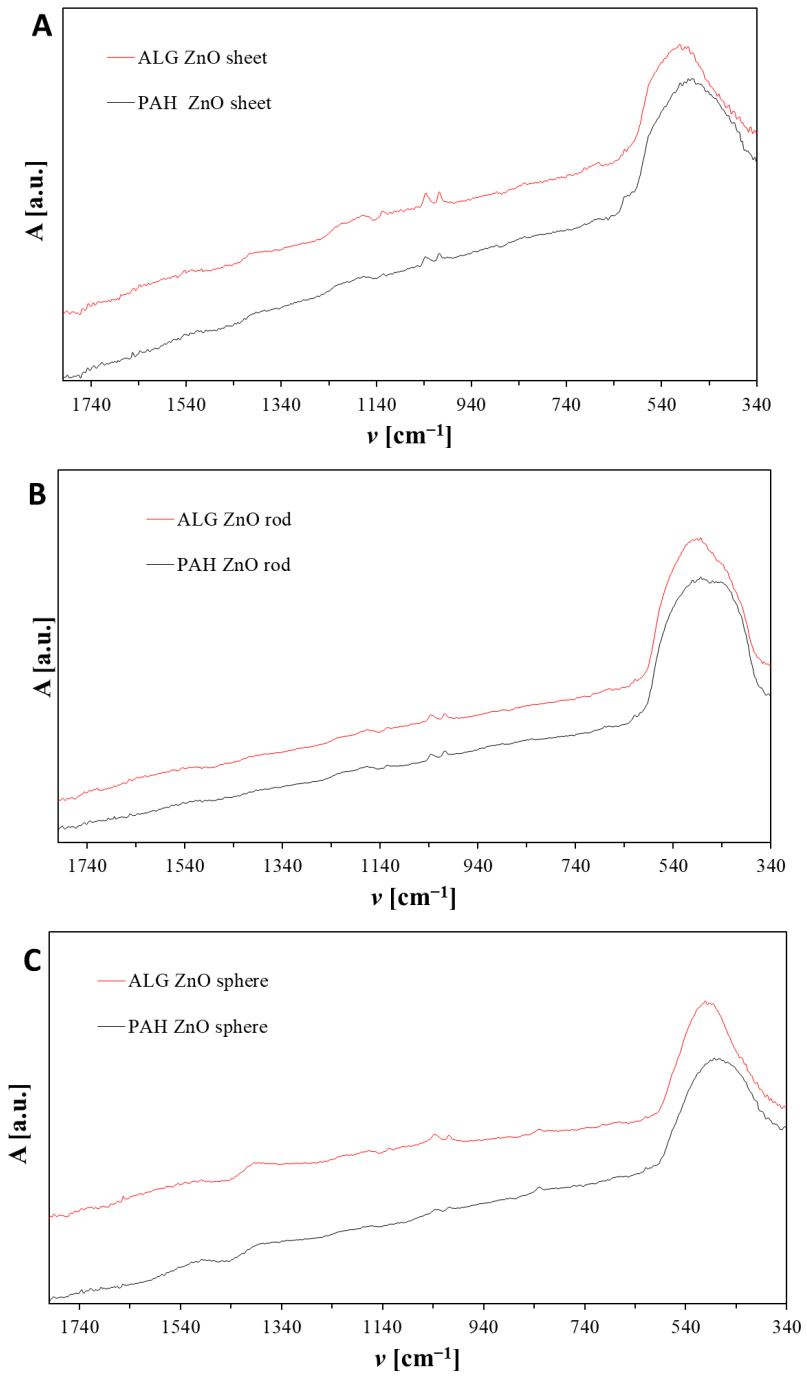

Figure S4. Fourier transformed infrared spectra showing the absorbance (A) at specific wavenumber (ν / cm^{-1}) for PAH and ALG terminating multilayers containing ZnO NPs exhibiting different morphology: (A) sheets, (B) rods and (C) spheres. PAH – poly(allylamine hydrochloride), ALG – alginate

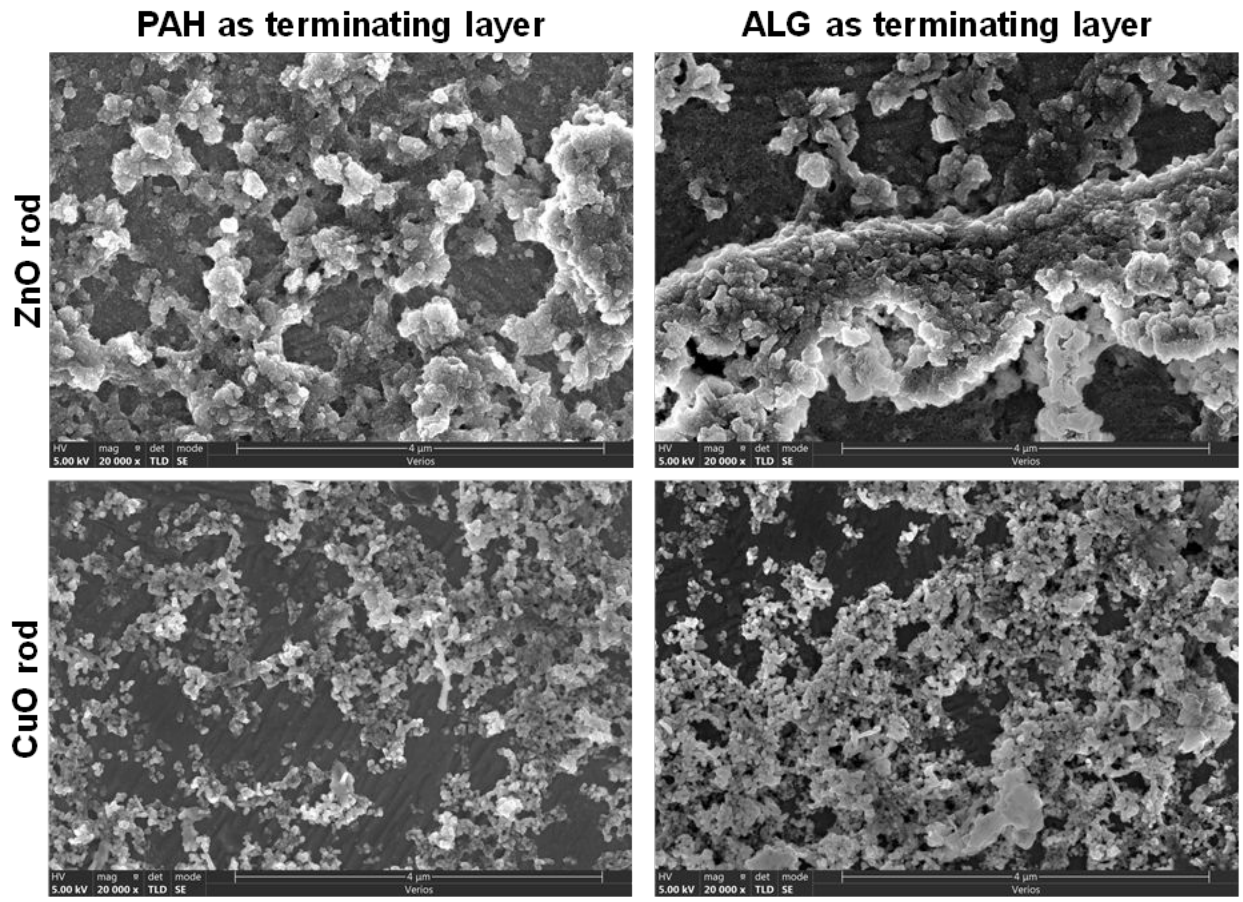

Figure S5. SEM micrographs of the stainless-steel surfaces after the antibacterial tests.